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regents prep chemistry: multiple-choice question archive - use in creating your own Regents-style Answer Number: 3. Explanation: Reduction occurs in oxidation-reduction reactions. Oxidation is a loss of

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oxidation-reduction reactions - purdue university - Oxidation-reduction reactions or redox reactions can involve the transfer of one or more electrons. Cu(s) Click here to check your answer to Practice Problem 2.

oxidation-reduction worksheet - leon m. goldstein high school - Answer questions on this sheet, I. Oxidation Numbers Are the following half reactions oxidation (ox.) or reduction (red)? a.

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balancing redox reactions : examples - chemwiki - GER = Gain Electron Reduction). Redox reactions usually and arrive at the answer above. By multiplying the oxidation half by 5 and the

oxidation- reduction reactions equations help? | - Apr 14, 2011 1. Which equation represents an oxidation-reduction reaction? (1) $Zn + 2HCl \rightarrow ZnCl_2 + H_2$ (2) $Zn(OH)_2 + 2HCl \rightarrow ZnCl_2 + 2H_2O$ (3) $H_2O + NH_3 \rightarrow NH_4^+$

oxidation and reduction reactions workbook - If the answer is no, write a balanced equation for the reaction that would occur. Also, state whether the reaction is oxidation or reduction.

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oxidation- reduction reactions - chemwiki: the - Oxidation-Reduction Reactions. Redox reactions are comprised of two parts, a reduced half and an oxidized half, that always occur together.

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